

Published on Web 08/04/2009

Catalytic Asymmetric Synthesis of Chiral Tertiary Organoboronic Esters through Conjugate Boration of β -Substituted Cyclic Enones

I-Hon Chen, Liang Yin, Wataru Itano, Motomu Kanai,* and Masakatsu Shibasaki*

Graduate School of Pharmaceutical Sciences, The University of Tokyo, Hongo, Bunkyo-ku, Tokyo113-0033, Japan

Received June 5, 2009; E-mail: mshibasa@mol.f.u-tokyo.ac.jp

Catalytic asymmetric conjugate addition reactions are a fundamental synthetic methodology to introduce functionalities at the β -position of carbonyl groups.¹ Despite the plethora of catalytic asymmetric conjugate addition reactions using β -monosubstituted α , β -unsaturated carbonyl substrates, methods applicable to the synthesis of β -tetrasubstituted carbonyl compounds using β , β disubstituted substrates are still limited.² This is partly due to the markedly lower reactivity of β , β -disubstituted substrates compared to β -monosubstituted substrates. In addition, chirality is difficult to control due to the smaller steric difference between the β -substituents. Enhanced catalyst activity and enantioselectivity are required to achieve catalytic asymmetric β -tetrasubstituted carbon construction.

To overcome these hurdles, we focused on Cu(I)-catalyzed conjugate boration using bis(pinacolato)diboron (2). The basic methodology was originally developed independently by Hosomi³ and Miyaura⁴ in racemic systems. The reaction proceeds through a borylcopper species generated via transmetalation.⁵ Yun recently identified the remarkable acceleration effects of protic additives in developing the first enantioselective version using linear β -monosubstituted α , β -unsaturated carboxylic acid derivatives as substrates and phosphines as ligands, 6a-d and subsequently Fernández and Guiry employed chiral NHC and P-N ligands, respectively.^{6e,f} Chiral organoboron compounds are versatile synthetic intermediates due to the convertibility of C-B bonds to a variety of functional groups.^{7,8} Furthermore, chiral organoboronic acids exhibiting unique biological activities were identified.9 Based on the reaction mechanism for an analogous Cu-catalyzed allylic boration proposed by Ito and Sawamura,¹⁰ chirality on the copper atom of the borylcopper species exists proximal to the prochiral carbon in the four-membered cyclic (Cu-B- β -C- α -C) transition state. This is advantageous for the strict enantio-differentiation required to overcome the abovementioned difficulty in enantio-selection. Here we report a catalytic enantioselective conjugate boration of β -substituted cyclic enones that produces enantiomerically enriched tertiary organoboronates.

We first optimized racemic reaction conditions using 3-phenyl-2-cyclohexen-1-one (**1a**) as a substrate and racemic BINAP as a ligand (Table 1, entries 1–3). Using a CuO'Bu complex generated in situ from CuCl and NaO'Bu in the presence of 2 equiv of MeOH in THF (Yun's conditions), the reaction did not proceed at all. On the other hand, product **3a** was obtained in 10% yield in DMF *in the absence of MeOH* (entry 1). After a systematic survey of copper sources and metal alkoxides, the optimized conditions were determined to be a combination of CuPF₆(CH₃CN)₄ and LiO'Bu; **3a** was obtained in 55% yield (entry 2). The detrimental effects of additive MeOH in this reaction were confirmed again at this point; yield decreased to 11% in the presence of 2 equiv of MeOH (entry 3).

The optimized conditions were next extended to an asymmetric version in the presence of chiral phosphines (entries 4-8). (*R*,*R*)-QuinoxP* containing a P-chirality¹¹ was the optimum chiral ligand;

	· _+	CuCl or CuPF ₆ ^e (10 mol%) O ligand (12 mol%) PinB-BPin ^t BuOM (15 mol%)					
Ph			DMF or DMSO [MeOH (2 equiv)]		B(Pin)		
1a		2 (1.5 equiv)	rt, 12 h		3a 🏾 🤇		
entry	М	ligand	solv.	MeOH	yield (%) ^b	ee (%) ^c	
1	Na	rac-BINAP	DMF	_	10	_	
2	Li	rac-BINAP	DMF	_	55	-	
3	Li	rac-BINAP	DMSO	+	11	-	
4	Li	tol-BINAP	DMF	_	82	48	
5	Li	DTBM-SEGPHOS	5 DMF	_	23	11	
6	Li	Ph-BPE	DMF	_	90	27	
7	Li	JOSIPHOS	DMF	_	55	20	
8	Li	QuinoxP*	DMF	_	78	91	
9	Li	QuinoxP*	DMSO	_	90	98	
10^{d}	Li	QuinoxP*	DMSO	_	88	98	
11^{d}	Na	QuinoxP*	DMSO	_	61	96	
12^{d}	Κ	QuinoxP*	DMSO	_	57	93	

Table 1. Optimization of Reaction Conditions

 a In entry 1, CuCl was used. In other entries, CuPF₆(CH₃CN)₄ was used. b Isolated yield. c Determined by chiral GC. d 5 mol % of catalyst.

excellent enantioselectivity (91% ee) was obtained with a yield of 78% (entry 8). DMSO was a better solvent than DMF, affording **3a** in 90% yield with 98% ee (entry 9). Catalyst loading was reduced to 5 mol % without affecting the enantioselectivity (entry 10). Interestingly, the enantioselectivity was not very sensitive to the hard metals of *t*-butoxides. The chemical yield, however, was significantly lower when using NaO'Bu and KO'Bu instead of LiO'Bu (entries 10-12).

The substrate scope was then investigated under the optimized conditions (Table 2). Generally high enantioselectivity (93–98% ee) was obtained from β -aromatic-substituted enones. Neither electron-donating nor withdrawing *para-* and *meta-*substitution on the β -aromatic ring affected the yield or enantioselectivity (entries 1–6). Methyl- and branched-alkyl substituted enones were competent substrates as well (entries 7–9). In addition to the 6-membered cyclic enones, 5- and 7-membered cyclic enones produced excellent to synthetically useful enantioselectivity (entries 10–12).

This reaction is a useful platform for the synthesis of various chiral building blocks that are otherwise difficult to access (Scheme 1). Oxidation of 3a with NaBO₃^{8b} afforded tertiary alcohol 4 in high yield with minimum loss of enantioselectivity (eq 1). Unique chiral tertiary organoboronic acid 5, which might be an interesting pharmaceutical lead, was produced through acid hydrolysis (eq 2). Unlike Yun's conditions,⁶ the present reaction did not require protic additives. Therefore, the products in the reaction mixtures are corresponding boron enolates that can react further with electrophiles. This advantage was illustrated by extension to a catalytic asymmetric three-component reaction using 1a, 2, and benzaldehyde (eq 3). Product 6, derived through enantioselective conjugate

Table 2.Catalytic Enantioselective Conjugate Boration of β -Substituted Cyclic Enones



entry	substrate	x (mol %)	yield (%) ^a	ee (%) ^b
1	R = Ph, n = 1 (1a)	5	88	98 ^d
2	$R = p-MeO-C_6H_4, n = 1$ (1b)	10	84	93
3	R = p-Me-C ₆ H ₄ , $n = 1$ (1c)	10	86	95
4	R = p-F-C ₆ H ₄ , $n = 1$ (1d)	5	80	93
5	R = m-MeO-C ₆ H ₄ , $n = 1$ (1e)	10	83	95
6	R = m-Me-C ₆ H ₄ , $n = 1$ (1f)	10	89	98
7^c	R = Me, n = 1 (1g)	5	91	81
8	$R = {}^{i}Pr, n = 1$ (1h)	10	91	94
9	$R = {}^{i}Bu, n = 1$ (1i)	10	92	85
10	R = Ph, n = 0 (1j)	10	85	98
11	$\mathbf{R} = {}^{i}\mathbf{B}\mathbf{u}, n = 0 (1\mathbf{k})$	10	94	70
12	R = Ph, n = 2 (11)	10	99	98

^{*a*} Isolated yield. ^{*b*} Determined by chiral GC or HPLC. ^{*c*} NaO'Bu was used instead of LiO'Bu. ^{*d*} Absolute configuration was assigned to be (R).¹³

Scheme 1. Synthetically Useful Conversions of the Products and Extension to a Catalytic Asymmetric Three-Component Reaction



boration, diastereoselective aldol reaction, and oxidation in one pot, was obtained in 71% yield with 91% ee (only two diastereomers were detected; dr = 6.5/1). Thus, the stereochemistries of three contiguous stereogenic centers involving a tetrasubstituted carbon were controlled in a high level, including their absolute configuration.¹²

Although it is not yet clear why protic additives are not necessary for this reaction, we currently believe that a catalytic amount of LiPF₆ generated in the active catalyst (CuO'Bu) formation step from CuPF₆ and LiO'Bu accelerates the probable turnover-limiting catalyst regeneration step.^{14,15} The acceleration effect of LiPF₆ was confirmed by comparison with a reaction using Li-free CuO'Bu¹⁶ catalyst (5 mol %); **3a** was obtained only in 31% yield with 75% ee (cf. Table 2, entry 1). The yield of **3a** improved to 80% (87% ee) when LiPF₆ (5 mol %) was added to this reaction mixture.

In conclusion, we developed an enantioselective conjugate boration of β -substituted cyclic enones to produce enantiomerically enriched tertiary boronates, catalyzed by a Cu-QuinoxP* complex. Catalytic LiPF₆ had positive effects on reactivity. Detailed mechanistic studies especially upon the role of LiPF₆ are currently ongoing. Acknowledgment. Financial support was provided by Grantin-Aid for Scientific Research (S) and (B) from JSPS. We thank T. Nitabaru and Y. Kimura for X-ray crystallographic studies of **3a**. I.H. thanks Iwaki Pharmaceutical Company for financial support.

Supporting Information Available: Experimental procedures and characterization of the products. This material is available free of charge via the Internet at http://pubs.acs.org.

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- JA9045839